INTERNATIONAL STANDARD

ISO 1000

Third edition 1992-11-01

SI units and recommendations for the use of their multiples and of certain other units

Unités SI et recommandations pour l'emploi de leurs multiples et de certaines autres unités.

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

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Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

International Standard ISO 1000 was prepared by Technical Committee ISO/TC 12, Quantities, units, symbols, conversion factors.

This third edition cancels and replaces the second edition (ISO 1000:1981). The major technical changes from the second edition are the following:

- the decision by the International Committee for Weights and Measures (Comité International des Poids et Mesures, CIPM) in 1980 concerning the status of supplementary units has been incorporated;
- quantities and units from ISO37, parts 9, 10, 12 and 13, have been added to annex A;
- the old definition of the metre in annex B has been replaced by the new definition.

The scope of Technical Committee ISO/TC 12 is standardization of units and symbols for quantities and units (and mathematical symbols) used within the different fields of science and technology, giving, where necessary, definitions of these quantities and units. Standard conversion factors for converting between the various units also come under the scope of the TC. In fulfilment of this responsibility, ISO/TC 12 has prepared ISO 31 and ISO 1000.

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Annex A forms an integral part of this International Standard. Annex B is for information only.

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SI units and recommendations for the use of their multiples and of certain other units

1 Scope

This International Standard

- a) describes the International System of Units (in clauses 3, 4 and 6);
- b) recommends selected decimal multiples and submultiples of SI units for general use and gives certain other units which may be used with the International System of Units (in clauses 5 and 7, and annex A);
- c) quotes the definitions of the SI base units (in annex B).

2 Normative reference

The following standard contains provisions which, through reference in this text, constitute provisions of this International Standard. At the time of publication, the edition indicated was valid. All standards are subject to revision, and parties to agreements based on this International Standard are encouraged to investigate the possibility of applying the most recent edition of the standard indicated below. Members of

EC and ISO maintain registers of currently valid International Standards.

IEC 27-1:1971,²⁾ Letter symbols to be used in electrical technology — Part 1: General

3 SI units

The name International System of Units (Système International d'Unités), with the international abbreviation SI, was adopted by the 11th General Conference on Weights and Measures (Conférence Générale des Poids et Mesures, CGPM) in 1960.

This system includes:

- base units
- derived units including supplementary units

which together form the coherent system of SI units.

3.1 Base units

The International System of Units is based on the seven base units listed in table 1.

¹⁾ Full information about the International System of Units is given in a publication by the International Bureau of Weights and Measures (Bureau International des Poids et Mesures, BIPM): Le Système International d'Unités (SI), including an authorized English translation.

^{2) 5}th edition, currently being revised.

Table 1 — SI base units

Dono muontitu	SI base unit			
Base quantity	Name	Symbol		
length	metre	m		
mass	kilogram	kg		
time	second	s		
electric current	ampere	А		
thermodynamic temperature	kelvin	К		
amount of substance	mole	mol		
luminous intensity	candela	cd		

For the definitions of the base units, see annex B.

3.2 Derived units including supplementary units

Derived units are expressed algebraically in terms of base units. Their symbols are obtained by means of the mathematical signs of multiplication and division; for example, the SI unit for velocity is metre per second (m/s).

For some of the SI derived units, special names and symbols exist; those approved by the CGPM are listed in tables 2 and 3.

The SI units radian and steradian are called supplementary units. They are "dimensionless" derived units (more precisely, derived units of dimension one) with special names and symbols. Although the coherent unit for plane angle and for solid angle is expressed by the number 1, it is convenient to use the special names radian (rad) and steradian (sr) respec-

tively instead of the number 1 in many practical cases; for example the SI unit for angular velocity can be written as radian per second (rad/s).

It may sometimes be useful to express derived units in terms of other derived units having special names; for example, the SI unit for electric dipole moment is usually expressed as $C \cdot m$ instead of $A \cdot s \cdot m$.

4 Multiples of SI units

The prefixes given in table 4 are used to form names and symbols of multiples (decimal multiples and submultiples) of the SI units.

The symbol of a prefix is considered to be combined with the kernel symbol³⁾ to which it is directly attached, forming with it a new symbol (for a decimal multiple or sub-multiple) which can be raised to a positive or negative power, and which can be combined with other unit symbols to form symbols for compound units.

EXAMPLES

1 cm³ =
$$(10^{-2} \text{ m})^3$$
 = 10^{-6} m^3
 $= \mu \text{s}^{-1}$ = $(10^{-6} \text{ s})^{-1}$ = 10^6 s^{-1}
1 mm²/s = $(10^{-3} \text{ m})^2/\text{s}$ = $10^{-6} \text{ m}^2/\text{s}$

Compound prefixes shall not be used; for example, write nm for nanometre, not $m\mu m$.

NOTE 1 For historical reasons the name of the base unit for mass, the kilogram, contains the name of the SI prefix "kilo". Names of the decimal multiples and sub-multiples of the unit of mass are formed by adding the prefixes to the word "gram", e.g. milligram (mg) instead of microkilogram (μ kg).

³⁾ In this case, the term "kernel symbol" means only a symbol for a base unit or a derived unit with a special name. See, however, note 1 in clause 4 about the base unit the kilogram.

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Table 2 — SI derived units with special names, including SI supplementary units

	With Special names,	SI derived unit				
Derived quantity	Special name	Symbol	Expressed in terms of SI base units and SI derived units			
plane angle	radian	rad	1 rad = 1 m/m = 1			
solid angle	steradian	sr	$1 \text{ sr} = 1 \text{ m}^2/\text{m}^2 = 1$			
frequency	hertz	Hz	$1 \text{ Hz} = 1 \text{ s}^{-1}$			
force	newton	N	$1 N = 1 kg \cdot m/s^2$			
pressure, stress	pascal	Pa	1 Pa = 1 N/m ²			
energy, work, quantity of heat	joule	J	1 J = 100 m			
power, radiant flux	watt	W	1 W = 1 J/s			
electric charge, quantity of electricity	coulomb	COOK	1 C = 1 A·s			
electric potential, potential difference, tension, electromotive force	volt forad . GWHO	IN PO.	1 V = 1 W/A			
capacitance	farad	F	1 F = 1 C/V			
electric resistance	ohm	Ω	1 Ω = 1 V/A			
electric conductance	siemens	s	$1 S = 1 \Omega^{-1}$			
magnetic flux	weber	Wb	1 Wb = 1 V · s			
magnetic flux density	tesla	Т	$1 T = 1 Wb/m^2$			
inductance	henry	Н	1 H = 1 Wb/A			
Celsius temperature	degree Celsius ¹⁾	°C	1 °C = 1 K			
luminous flux	lumen	lm	1 lm = 1 cd · sr			
illuminance	lux	lx	$1 \text{ lx} = 1 \text{ lm/m}^2$			

¹⁾ Degree Celsius is a special name for the unit kelvin for use in stating values of Celsius temperature. (See also note 6 concerning the kelvin in annex B.)

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Table 3 — SI derived units with special names admitted for reasons of safeguarding human health

	SI derived unit					
Derived quantity	Special name	Symbol	Expressed in terms of SI base units and SI derived units			
activity (of a radionuclide)	becquerel	Bq	$1 \text{ Bq} = 1 \text{ s}^{-1}$			
absorbed dose, specific energy imparted, kerma, absorbed dose index	gray	Gy	1 Gy = 1 J/kg			
dose equivalent, dose equivalent index	sievert	Sv	1 Sv = 1 J/kg			

Table 4 — SI prefixes

_	Pre	fix		
Factor	Name	Symbol		
10 ²⁴ 10 ²¹ 10 ¹⁸ 10 ¹⁵	yotta zetta exa peta	Y Z E P		
10 ¹² 10 ⁹ 10 ⁶ 10 ³	tera giga mega kilo	T G M k		
10 ² 10 10 ⁻¹ 10 ⁻²	hecto deca deci centi	h da d		
10 ⁻³ 10 ⁻⁶ 10 ⁻⁹ 10 ⁻¹²	milli micro nano pico	n n		
10 ⁻¹⁵ 10 ⁻¹⁸ 10 ⁻²¹ 10 ⁻²⁴	femto atto zepto yocto	f a z y		

5 Use of SI units and their multiples

5.1 The choice of the appropriate multiple (decimal multiple or sub-multiple) of an SI unit is governed by convenience, the multiple chosen for a particular application being the one which will lead to numerical values within a practical range.

5.2 The multiple can usually be chosen so that the numerical values will be between 0,1 and 1 000. In the case of a compound unit containing a unit to the second or third power, this is not always possible.

EXAMPLES

1,2 × 10 ⁴ N 0,003 94 m	can be written as	12 kN
0,003 94 m	can be written as	3,94 mm
13401 Pa	can be written as	1,401 kPa
3.1×10^{-8} s	can be written as	31 ns

However, in a table of values of the same quantity or in a discussion of such values within a given context, it will generally be better to use the same multiple for all items, even if some of the numerical values will then be outside the range 0,1 to 1 000. For certain quantities in particular applications, the same multiple is customarily used; for example, the millimetre is used for dimensions in most mechanical engineering drawings.

- **5.3** The number of prefixes used in forming compound units should be limited as far as is compatible with practical usage.
- **5.4** Errors in calculations can be avoided more easily if all quantities are expressed in SI units, powers of 10 being used instead of prefixes.

6 Rules for writing unit symbols

6.1 Unit symbols shall be printed in roman (upright) type (irrespective of the type used in the rest of the text), shall remain unaltered in the plural, shall be written without a final full stop (period) except for normal punctuation, e.g. at the end of a sentence, and shall be placed after the complete numerical value in

the expression for a quantity, leaving a space between the numerical value and the unit symbol.

Unit symbols shall in general be written in lower case letters except that the first letter is written in upper case when the name of the unit is derived from a proper name.

EXAMPLES

m	metre

s second

A ampere

Wb weber

6.2 When a compound unit is formed by multiplication of two or more units, this should be indicated in one of the following ways:

NOTES

- 2 In systems with limited character sets a dot on the line is used instead of a half-high dot.
- 3 The latter form may also be written without a space, provided that special care is taken when the symbol for one of the units is the same as the symbol for a prefix, e.g. mN is used only for millinewton, not for metre newton.

When a compound unit is formed by dividing one unit by another, this should be indicated in one of the following ways:

$$\frac{m}{s}$$
, m/s $m \cdot s^{-1}$

A solidus (/) shall not be followed by a multiplication sign or a division sign on the same line unless parentheses are inserted to avoid any ambiguity. In complicated cases negative powers or parentheses shall be used.

7 Non-SI units which may be used with SI units and their multiples

- **7.1** There are certain units, outside the SI, recognized by the CIPM as having to be retained because of their practical importance (see tables 5 and 6).
- **7.2** Prefixes given in table 4 may be attached to some of the units given in tables 5 and 6; for example, millilitre, ml. (See also annex A, column 6.)
- 7.3 In a limited number of cases compound units are formed with the units given in tables 5 and 6 together with SI units and their multiples; for example, kg/h; km/h. (See also annex A, columns 5 and 6.)
- NOTE 4 There are some other units outside the SI which are recognized by the CIPM for temporary use. They are given in column 7 of the table in annex A and marked by an asterisk (*).

Table 5 — Units used with the SI

e a	Unit				
Quantity	Name	Symbol	Definition		
time	minute	min	1 min = 60 s		
	hour	h	1 h = 60 min		
	day	d	1 d = 24 h		
plane angle	degree	۰	$1^{\circ} = (\pi/180) \text{ rad}$		
	minute	,	1' = (1/60)°		
	second	"	1'' = (1/60)'		
volume	litre	l, L1)	$1 I = 1 dm^3$		
mass	tonne ²⁾	t	$1 t = 10^3 kg$		

- 1) The two symbols for the litre are on an equal footing. The CIPM will, however, make a survey on the development of the use of the two symbols in order to see if one of the two may be suppressed.
- 2) Also called the metric ton in the English language.

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Table 6 — Units used with the SI, whose values in SI units are obtained experimentally

0	Unit					
Quantity	Name	Symbol	Definition			
energy	electronvolt	eV	The electronvolt is the kinetic energy acquired by an electron in passing through a potential difference of 1 volt in vacuum: 1 eV ≈ 1,602 177 × 10 ⁻¹⁹ J.			
mass	unified atomic mass unit	u	The unified atomic mass unit is equal to 1/12 of the mass of an atom of the nuclide 12 C: 1 u \approx 1,660 540 \times 10 ⁻²⁷ kg.			

Annex A

(normative)

Examples of decimal multiples and sub-multiples of SI units and of some other units which may be used

For a number of commonly used quantities, examples of decimal multiples and sub-multiples of SI units, as well as of some other units which may be used, are given in this annex. It is suggested that the selection shown, while not intended to be restrictive, will none the less prove helpful in presenting values of quantities in an identical manner in similar contexts within the various sectors of technology. For some needs (for example, in applications in science and education), it is recognized that greater freedom will be required in the choice of decimal multiples and sub-multiples of SI units than is exemplified in the list which follows.

Item No. in ISO 31: 1992	Quantity	SI unit	Selection of multiples and sub- multiples of the SI unit	by the CIPM retained, and some of their	the SI recognized as having to be for special cases combinations with units Multiples or sub-multiples of units given in column 5	Remarks and information about units used in special fields
1	2	3	4	(6)	6	7
Part 1: Spa	ace and time		<u> </u>	<u> </u>		
1-1	angle, (plane angle)	rad (radian)	Mrad µrad	(degree) $1^{\circ} = \frac{\pi}{180} \text{ rad}$ (minute) $1' = \frac{1^{\circ}}{60}$ (second) $1'' = \frac{1'}{60}$		gon [gon (or grade)], $1 \text{ gon} = \frac{\pi}{200} \text{ rad}$ If the radian is not used, the units degree or gon (or grade) may be used. Decimal subdivisions of degree are preferable to minute and second for most applications. For the units degree, minute and second for plane angle, there shall be no space between the numerical value and the unit symbol.
1-2	solid angle	sr (steradian)				
1-3.1	length	m (metre)	km cm mm µm nm pm fm			1 nautical mile* = 1 852 m (exactly) * Recognized by the CIPM for temporary use.

Item No.	Quantity	of multiples SI unit and sub-	Selection of retained, and for special cases some of their combinations with	retained, and for special cases some of their combinations with SI units		Remarks and information about
1992			multiples of the SI unit	Units	Multiples or sub-multiples of units given in column 5	units used in special fields
1	2	3	4	5	6	7
1-4	area	m²	km ² dm ² cm ² mm ²			ha* (hectare), 1 ha = 10 ⁴ m ² a* (are), 1 a = 10 ² m ² * Recognized by the CIPM for temporary use.
1-5	volume	m ³	dm³	I, L (litre) 1 I = 10 ⁻³ m ³ = 1 dm ³	hI 1 hI = 10^{-1} m ³ cl 1 cl = 10^{-5} m ³ mI	In 1964, the CGPM declared that the name litre (I) may be used as a special name for the cubic decimetre (dm³) and advised against the use of the name litre for high-precision measurements. See also footnote 1) to table 5.
1-7	time		mm ³	d (day)	$1 \text{ m}^3 = 10^{-6} \text{ m}^3 $	Other units such as week, month and year (a) are in common use.
		ARDSIS	ks ks	1 d= 24 h (exactly) h (hour) 1 h = 60 min (exactly) min (minute) 1 min = 60 s (exactly)		The definitions of month and year often need to be specified.
	STAND	(second)	ms µS ns			
1-8	angular velocity	rad/s				
1-10	velocity	m/s		m/h	km/h 1 $km/h = \frac{1}{3.6}$ m/s	1 knot* = 1,852 km/h (exactly) = 0,514 444 m/s For the hour, see item No. 1-7. * Recognized by the CIPM for temporary use.
1-11.1	acceleration	m/s ²				

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Item No.	Quantity	SI unit	Selection of multiples and sub-	by the CIPM retained, and some of their	the SI recognized I as having to be I for special cases combinations with I units	Remarks and information about
1992	·		multiples of the SI unit	Units	Multiples or sub-multiples of units given in column 5	units used in special fields
1	2	3	4	5	6	7
Part 2: Per	iodic and related	d phenomen	a		,	T
2-3.1	frequency	Hz (hertz)	THz GHz MHz kHz			100.1095
2-3.2	rotational frequency	s ⁻¹		min ⁻¹	UII POF OF IS	The designations "revolutions per minute" (r/min) and "revolutions per second" (r/s) are widely used for rotational frequency in specifications on rotating machinery. (See also IEC 27-1.) For the minute, see item 1-7.
2-4	angular frequency	rad/s		an the		
Part 3: Me	echanics			jie		
3-1	mass	kg (kilogram)	Mg g mg µg	(tonne) 1 t = 10 ³ kg		See footnote 2) to table 5.
3-2	volumic mass, density, mass density	kg/m³	Mg/m ³ or kg/dm ³ or g/cm ³	t/m ³ or kg/l	g/ml g/l	For the litre, see item No. 1-6. For the tonne, see item No. 3-1.
3-5	lineic mass,	kg/m				$1 \text{ tex} = 10^{-6} \text{ kg/m} = 1 \text{ g/km}$
C	linear density		mg/m			The unit tex is used for textile filaments.
3-7	moment of inertia	kg·m²				
3-8	momentum	kg·m/s				
3-9.1	force	N (newton)	MN kN mN μN			

Item No.	Quantity	SI unit	Selection of multiples and sub-	by the CIPM retained, and some of their	the SI recognized las having to be for special cases combinations with lunits	Remarks and information about units used in special fields
1992			multiples of the SI unit	Units	Multiples or sub-multiples of units given in column 5	units used in special neids
1	2	3	4	5	6	7
3-11	moment of momentum, angular momentum	kg·m²/s				CO 100.1092
3-12.1	moment of force	N⋅m	MN·m kN·m mN·m μN·m			, 150 1000.
3-15.1	pressure	Pa (pascal)	GPa MPa kPa hPa mPa µPa	in	withefull POK	bar* (bar), 1 bar = 100 kPa (exactly) 1 mbar = 1 hPa The use of the bar shall be restricted to existing uses in the field of fluid pressure. * Recognized by the CIPM for temporary use.
3-15.2	normal stress	Pa	GPa MPa kPa	Clickto		
3-23	viscosity, (dynamic viscosity)	Pa·s	mPa-s			P (poise) 1 cP = 1 mPa·s The poise and stokes are special names for CGS units. They and their multiples and sub-multiples shall not be used together with SI units.
3-24	kinematic viscosity	m²/s	mm²/s			St (stokes) 1 cSt = 1 mm ² /s See remark on item 3-23.
3-25	surface tension	N/m	mN/m			
3-26.1 and 3-26.2	energy, work	J (joule)	EJ PJ TJ GJ MJ kJ			

Item No. in ISO 31:	Quantity	SI unit	Selection of multiples and sub- multiples of the SI unit	by the CIPN retained, and some of their	the SI recognized I as having to be I for special cases combinations with	Remarks and information about
1992				Units	Multiples or sub-multiples of units given in column 5	units used in special fields
1	2	3	4	5	6	7
3-27	power	W (watt)	GW MW kW mW			
Part 4: He	at	L	L.:			
4-1	thermo- dynamic temperature	K (kelvin)			ak of le	
4-2	Celsius temperature	°C (degree Celsius)		to liew the	FUII PO	The Celsius temperature t is equal to the difference $(T-T_0)$ between two thermodynamic temperatures T and T_0 , where $T_0 = 273,15$ K (exactly). For the definition and the use of the degree Celsius (°C), see
			NL.	60		note 6 under the definition of the kelvin in annex B.
4-3.1	linear expansion coefficient	K ⁻¹	V. Clica			For the degree Celsius, see item 4-2.
4-6	heat	J	EJ PJ TJ GJ MJ kJ mJ			
4-7	heat flow rate		kW			
(b`	W				
4-9	thermal conductivity	W/(m·K)				For the degree Celsius, see item 4-2.
4-10.1	coefficient of heat transfer	W/(m ² · K)				For the degree Celsius, see item 4-2.
4-11	thermal insulance	m² · K/W				For the degree Celsius, see item 4-2.
4-15	heat capacity	J/K	kJ/K			For the degree Celsius, see item 4-2.

Item No. in ISO 31: 1992	Quantity	SI unit	Selection of multiples and sub-	by the CIPM retained, and some of their o	the SI recognized as having to be for special cases combinations with units	Remarks and information about units used in special fields
	·		multiples of the SI unit	Units	Multiples or sub-multiples of units given in column 5	units used in special neros
1	2	3	4	5	6	7
4-16.1	massic heat capacity	J/(kg·K)	kJ/(kg⋅K)			For the degree Celsius, see item 4-2.
4-18	entropy	J/K	kJ/K			For the degree Celsius, see item 4-2.
4-19	massic entropy	J/(kg·K)	kJ/(kg ⋅ K)			For the degree Celsius, see item 4-2.
4-21.2	massic thermo- dynamic energy	J/kg	MJ/kg kJ/kg		ook	otis
Part 5: Ele	ectricity and mag	netism				
5-1	electric current	A (ampere)	kA mA μA nA pA	ct to le	Hitele	
5-2	electric charge, quantity of electricity	C (coulomb)	kCON µC nC pC	A h 1 A · h = 3,6 kC		For the hour, see item No. 1-7.
5-3	volumic charge, volume density of charge, charge density	C/m³	C/mm ³ or GC/m ³ MC/m ³ or C/cm ³ kC/m ³ mC/m ³			
5-4	areic charge, surface density of charge	C/m²	MC/m ² or C/mm ² C/cm ² kC/m ² mC/m ²			

Item No. in ISO 31: 1992	Quantity	SI unit	Selection of multiples and sub-	by the CIPN retained, and some of their	the SI recognized I as having to be I for special cases combinations with I units	Remarks and information about units used in special fields
			multiples of the SI unit	Units	Multiples or sub-multiples of units given in column 5	units used in special fields
1	2	3	4	5	6	7
5-5	electric field strength	V/m	MV/m kV/m or V/mm V/cm mV/m µV/m			200.100gr
5-6.1	electric potential	V	MV kV		C	
5-6.2	potential difference, tension	(volt)	mV μV		Full PDF of IS	
5-6.3	electromotive force			0	FULL PROPERTY.	
5-7	electric flux density	C/m ²	C/cm ² kC/m ² mC/m ² µC/m ²	Oilenite		
5-8	electric flux	С	MC KC			
5-9	capacitance	F (farad)	mF μF nF pF			
5-10.1	permittivity	F/m	μF/m nF/m pF/m			
5-13	electric polarization	C/m²	C/cm ² kC/m ² mC/m ² µC/m ²			
5-14	electric dipole moment	C·m				

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1992			multiples of the SI unit	Units	Multiples or sub-multiples of units given in column 5	units used in special fields
1	2	3	4	5	6	7
5-15	areic electric current, electric current density	A/m²	MA/m ² or A/mm ² A/cm ² kA/m ²			0.70gr
5-16	lineic electric current, linear electric current density	A/m	kA/m or A/mm A/cm		~	ot 150 100:1092
5-17	magnetic field strength	A/m	kA/m or A/mm A/cm		Withe full PDF	
5-18.1	magnetic potential difference	А	kA mA	jie	Mille	
5-19	magnetic flux density, magnetic induction	T (tesla)	mT μT nT	Click to V.		
5-20	magnetic flux	Wb (weber)	mWb			
5-21	magnetic vector potential	Wb/m	kWb/m or Wb/mm			
5-22.1	self- inductance	H (henry)	mH			
5-22.2	mutual inductance		μΗ nH pH			
5-24	permeability	H/m	μH/m nH/m			
5-27	magnetic moment, electro- magnetic moment	A·m²				

Item No. in ISO 31:	Quantity	SI unit		by the CIPM retained, and some of their	the SI recognized I as having to be I for special cases combinations with I units	Remarks and information about units used in special fields
1992			multiples of the SI unit	Units	Multiples or sub-multiples of units given in column 5	units used in special nerds
1	2	3	4	5	6	7
5-28	magnetization	A/m	kA/m or A/mm			℃
5-29	magnetic polarization	Т	mT			0.703
(IEC 27-1:1971, item 86)	magnetic dipole moment	N · m²/A or Wb · m			, <2	0/00
5-33	resistance (to direct current)	Ω (ohm)	GΩ MΩ kΩ mΩ μΩ	in	Full POF OF IS	
5-34	conductance (of direct current)	S (siemens)	kS mS μS	ojen		
5-36	resistivity	Ω·m .Ο	$G\Omega \cdot m$ $M\Omega \cdot m$ $k\Omega \cdot m$ $\Omega \cdot cm$ $m\Omega \cdot m$ $\mu\Omega \cdot m$ $n\Omega \cdot m$			$\frac{\Omega \cdot mm^2}{m} \left(= 10^{-6} \Omega \cdot m = \mu\Omega \cdot m \right)$ is also used.
5-37	conductivity	S/m	MS/m kS/m			
5-38	reluctance	H ⁻¹				
5-39	permeance	Н				
5-44.1	impedance, (complex impedance)	Ω	MΩ kΩ mΩ			
5-44.2	modulus of impedance, (impedance)		111000			
5-44.3	resistance					
5-44.4	reactance					

Item No. in ISO 31: 1992	Quantity	SI unit	Selection of multiples and sub- multiples of the SI unit	by the CIPM retained, and some of their	the SI recognized as having to be for special cases combinations with units	Remarks and information about units used in special fields
	-			Units	Multiples or sub-multiples of units given in column 5	units used in special neids
1	2	3	4	5	6	7
5-45.1	admittance, (complex admittance)	S	kS mS μS			97
5-45.2	modulus of admittance, (admittance)		,			20 100:1092
5-45.3	conductance					,50
5-45.4	susceptance					Š
5-49	active power	W	TW GW MW kW mW µW		Athe full POF	In electric power technology, active power is expressed in watts (W), apparent power in volt amperes (V · A) and reactive power in vars (var).
5-52	active energy	20515C	kJ GJ	W·h 1 W·h = 3,6 kJ (exactly)	TW·h GW·h MW·h kW·h	For the hour, see item 1-7.
Part 6: Lig	ht and related	ectromagne	etic radiatio	ns		
6-3	wavelength	m	μm nm pm			\mathring{A} * (ångström), 1 \mathring{A} = 10 ⁻¹⁰ m = 10 ⁻¹ nm = 10 ⁻⁴ μ m * Recognized by the CIPM for temporary use.
6-7	radiant energy	J				
6-10	radiant power, radiant energy flux	W				
6-13	radiant intensity	W/sr				

Item No. in ISO 31: 1992	Quantity	SI unit	Selection of multiples and sub-	by the CIPM retained, and some of their	the SI recognized las having to be for special cases combinations with lunits	Remarks and information about units used in special fields
			multiples of the SI unit	Units	Multiples or sub-multiples of units given in column 5	units used in special neids
1	2	3	4	5	6	7
6-14	radiance	W/(sr · m²)				
6-15	radiant exitance	W/m²				7097
6-16	irradiance	W/m²				200.
6-29	luminous intensity	cd (candela)			C	9
6-30	luminous flux	lm (lumen)			401	
6-31	quantity of light			Im · h 1 Im · h = 3 600 Im · s (exactly)	Full PO	For the hour, see item 1-7.
		lm·s		11/1		
6-32	luminance	cd/m²		ijew		
6-33	luminous exitance	lm/m²	٢٠.،	0		
6-34	illuminance	lx (lux)	V. Cy			
6-35	light exposure	lx · s	7			
6-36.1	luminous efficacy	1m/JV				
Part 7: Ac	oustics	L	L	J	Marie Control	<u> </u>
7-1	period periodic time	s	ms μs			
7-2	frequency	Hz	MHz kHz			
7-5	wavelength	m	mm			
7-8	volumic mass, mass density, density	kg/m ³				

Item No. in ISO 31: 1992	Quantity	SI unit	Selection of multiples and sub-	by the CIPM retained, and some of their	the SI recognized as having to be for special cases combinations with units	Remarks and information about units used in special fields
			multiples of the SI unit	Units	Multiples or sub-multiples of units given in column 5	units used in special neids
1	2	3	4	5	6	7
7-9.1 7-9.2	static pressure	Pa	mPa μPa			32
7-9.2	(instantaneous) sound pressure					00.703
7-11	(instantaneous) sound particle velocity	m/s	mm/s		L.	of 150 1000:1092
7-13	(instantaneous) volume flow rate	m³/s			FULLE	
7-14.1	velocity of sound	m/s			100	
7-16	sound power	W	kW mW μW pW	jick to je		
7-17	sound intensity	W/m²	mW)m² µW/m² pW/m²			
7-18	acoustic impedance	Pa·s/m³				
7-19	mechanical impedance	N · s/m				
7-20.1	surface density of mechanical impedance	Pa·s/m				
7-21	sound pressure level					B (bel) dB (decibel), 1 dB = 10^{-1} B
7-22	sound power level					B dB